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Answer Key

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Solubility and

Temperature Launch

Gizmo Add varying
amounts of a chemical
to a beaker of water to
create a solution,
observe that the
chemical dissolves in
the water at first, and

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then measure the concentration of the solution at the saturation point.

Answer Key

Solubility and Temperature Gizmo : Lesson Info ...

Add varying amounts of a chemical to a beaker of water to create a solution, observe that the chemical dissolves in the water at first, and then measure the concentration of the

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Exploration

solution at the saturation point. Either potassium nitrate or sodium chloride can be added to the water, and the temperature of the water can be adjusted.

Solubility and Temperature Gizmo : ExploreLearning

2019 Name: _____

Date: _____ Student

Exploration: Solubility and Temperature

Vocabulary:

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Exploration

concentration,
dissolve, homogeneous
mixture, solubility,
solubility curve, solute,
solution, solvent

Prior Knowledge Questions

(Do these BEFORE

using the Gizmo.) 1.

What happens when
you stir a spoonful of
sugar into hot water?

Solubility &

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GIZMO-1 (2) (1).pdf -

Name Date ...

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Solubility and
Temperature.

Vocabulary:
concentration,

dissolve, homogeneous
mixture, solubility,

solubility curve, solute,
solution, solvent. Prior

Knowledge Questions
(Do these BEFORE

using the Gizmo.). 1.

What happens when
you stir a spoonful of
sugar into hot water?

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(answers ...

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Solubility and
Temperature. Add
varying amounts of a
chemical to a beaker of
water to create a
solution, observe that
the chemical dissolves
in the water at first,

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Temperature ...
and then measure the
concentration of the
solution at the
saturation point. ...

Answer Key

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Exploration

Add varying amounts of a chemical to a beaker of water to create a solution, observe that the chemical dissolves in the water at first, and then measure

Student Exploration Solubility And Temperature Gizmo

...

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Vocabulary:

concentration,
dissolve, homogeneous
mixture, solubility,
solubility curve, solute,
solution, solvent Prior
Knowledge Questions

- (Do these BEFORE using the Gizmo.)
1. What happens when you stir a spoonful of sugar into hot water? The sugar will dissolve.
 2. When sugar or another substance is dissolved ...

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Vocabulary:
concentration,
dissolve, homogeneous
mixture, solubility,
solubility curve, solute,
solution, solvent
Prior Knowledge Questions
(Do these BEFORE
using the Gizmo.) 1.

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What happens when you stir a spoonful of sugar into hot water?

**Student Exploration:
Solubility and
Temperature**

Practical experiment investigating how temperature affects the solubility of different solid solutes (Junior Chemistry). Full prac report format, including discussion questions. Includes detailed teacher notes.

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**Temperature and
Solubility [Practical]
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Exploration Solubility

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develops online

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associated Realtime Reporting System are protected by US Patent Page 2/5. Explore Learning Solubility And Temperature Answer ...

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Answers**

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Sheet Solubility And
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It's easier to dissolve

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soluble solutions in

warm water than in

cool water. In this

science worksheet,

your child learns how

heat affects solubility

and plots data from a

table to create a line

graph. Student

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Answers

The rate of solubility,

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the amount of a solute that will dissolve in a solvent in a given period of time, is dependent upon chemical composition, quantity, and temperature. Plan your 60-minute lesson in Science or Chemistry with helpful tips from Keith Wright

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solubility temperature

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gizmo or just about any type of ebooks, for any type of product.

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Answer Key Add

varying amounts of a chemical to a beaker of water to create a solution, observe that the chemical dissolves in Page 13/29.

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Gizmo Answers For

Solubility And

Gizmo Answers For Solubility And

Page 25/29

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Temperature

The Temperature and Particle Motion

Gizmo™ illustrates how the molecules of gas

move at different

temperatures. In this

Gizmo, temperature is measured on the

Kelvin scale, which

measures temperature

from absolute zero, the

coldest possible

temperature (-273.15

°C). Each unit on the

Kelvin scale is

equivalent to 1 °C:

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273.15 K = 0 °C, and

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