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Summary. • Radioactive decay process governed by simple first order rate law: -dN/dt = λ tot N(t) • Decay rate is inversely proportional to half-life: λ tot = 0.693/T1/2. • Decay rates for one isotope are additive: λ tot = $\lambda \alpha + \lambda \beta + ...$ • Half-lives are not additive. • Decay chains can be modeled like a series of waterfalls.

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Solution: The hertz is used for periodic phenomena and equals the number of "cycles per second." The curie is used for the random or stochastic rate at which a radioactive source decays, specifically, 1 Ci = 3.7 × 1010 decays/second. 2. Advantages of SI units are apparent when one is presented with units of barrels, ounces, tons, and many others.

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Fundamentals of Nuclear Science and Engineering, 3rd edition. by J. Kenneth Shultis and Richard E. Faw. CRC Press, Boaca Raton, 2017. ISBN 978-1-4987-6929-7

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