

# Chemistry Nail Lab Answers

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### **Chemistry Nail Lab Answers**

Modeling Chemistry 1 U7 Nail Lab 2014 Name Date Pd Chemistry Unit 7 - Nail Lab Purpose The purpose of this investigation is to qualitatively and quantitatively describe the chemical reaction between copper(II) chloride and iron. Procedure Day 1 (Tues or Wed) 1. Label, then mass a plastic cup provided by the instructor. 2.

### **Date Pd Chemistry Unit 7 Nail Lab**

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### **Chemistry Nail Lab Answers - Under Book**

Hot Nail Lab, Daniel R. Barnes, Chemistry, 11/22/04 10:39:55 AM  
page 2 POST-LAB: 1.  $T = T_f - T_i$   $Q = m T C_p$   $C_p(H_2O) = 4.18 \text{ J/g oC}$   $C_p(Fe) = 0.449 \text{ J/g oC}$  Using the formulas and constants shown above, along with the appropriate data from your data table, perform the

### **Hot nail lab DRB - Hawthorne High School**

Alexis Davidson TA: Christian Gervasi 10/23/12 Galvanized Nails, Quality Control and an Introduction To Green Chemistry  
Galvanized Nails Reaction with Hydrochloric Acid Introduction  
Galvanized nails are nails that have been coated in zinc to protect the actual nail from rusting. Even though the zinc does not cover the nail forever, it allows the process of the rusting to be delayed; therefore ...

### **Galvanized Nails lab 4 - Alexis Davidson TA Christian ...**

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Crockett Maggie Crockett Brett Duell October 12, 2015  
Galvanized Nails Galvanized Nails, Quality Control, and an Introduction to Green Chemistry Introduction Nails are used everyday; from building monumental to minor projects, their quality is crucial. Last November, in Spokane, Washington there were record breaking rainstorms of up to 1.9 inches.

### **Galvanized nail lab report - Crockett 1 MaggieCrockett ...**

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### **Chemistry Nail Lab Answers - [happybabies.co.za](http://happybabies.co.za)**

So, we did a lab in class called Nail Lab. We had to get three iron nails and put them into copper (ii) chloride solution. They had

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instant rust. We measured and all. But my questions are the conclusion questions: 1) Why did the reaction stop? Which reactant was used up? How do you know? 2) Describe what was happening to the atoms of iron and copper during the reactions.

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### **Unit 7 Nail Lab Chemistry Conclusion Answers**

Chemistry Nail Lab Conclusion Answers - Maharashtra€Get Free Unit 7 Nail Lab Chemistry Conclusion Answers this lab was to observe the amount copper produced and iron used in a reaction

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and then produce the ratio of the copper atoms to the iron atoms. Materials: 1. Electronic balance 2. 250 ml beaker 3. 100 ml Chemistry Nail Lab Answers -

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...

Hi I'm working on a lab report called "Stoichiometry Lab: Hard as Nails?" I need to show my calculations for the mass and moles of iron that reacted, the mass and moles of copper that was produced. The reaction is  $\text{Fe}(s) + \text{CuCl}_2(aq) = \text{FeCl}_2(aq) + \text{Cu}(s)$ . Mass of nails pre-reaction 5.57g. Mass of nails post reaction 3.29g. Mass of product (Copper) 2.69g.

### **Chemistry Lab report help? | Yahoo Answers**

Iron Nail Rusting Chemistry Lab Help Please? What factors are known to speed up or slow down corrosion (rusting) of iron? If I was to experiment 5 different iron nails, what would be some

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good ideas as to getting good experimental data that will show which factors speed up the process of rusting

### **Iron Nail Rusting Chemistry Lab Help Please? | Yahoo Answers**

(Your answer is the hypothesis) Objective: Students will be able to determine if a reaction is a single displacement and how to predict the products of a single replacement reaction. Procedure : DAY ONE 1. Use the steel wool to clean the nail. You want the nail to be as shiny as possible. 2. Mass the nail and record the mass in the data section. 3.

### **3. Use the balance to measure the mass the nail and record ...**

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We dropped iron nails in copper chloride with water solution.

- 1.what are any precautions you need to take?
- 2.explain what and aqueous solution is and give an example from the lab?
- 3.when you decanted identify the solution and the precipitate?
- 4.why did you rinse the precipitate with water then hydrochloric acid and again with water?
- 5.please write the balanced chemical equation for this ...

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This video shows the setup of the long term reaction of  $\text{CuCl}_2$  and an iron nail. [www.dlt.ncssm.edu](http://www.dlt.ncssm.edu) Please attribute this work as being created by the North Ca...

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